

## POLYACETYLENES FROM *PANAX GINSENG* ROOTS

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**Key Word Index**—*Panax ginseng*; Araliaceae; roots;  $C_{17}$  acetylenes; heptadeca-1-ene-4,6-diyne-3,9,10-triol.

**Abstract**—Two major and two minor polyacetylenes were isolated from fresh Korean ginseng roots. The chemical structure of the two major polyacetylenes separated were determined by UV, IR,  $^1H$  NMR,  $^{13}C$  NMR, mass spectra and elemental analysis. One of them was a new compound identified as heptadeca-1-ene-4,6-diyne-3,9,10-triol.

### INTRODUCTION

A polyacetylene compound, panaxynol, from *Panax ginseng* roots was isolated by Takahashi *et al.* in 1964 [1, 2]. The chemical structure of the compound was identical to faltarinol [3] isolated from *Falcaria vulgaris* and carotatoxin isolated from *Daucus carota* [4]. Wrobel *et al.* [5] in 1973 also isolated another type of  $C_{17}$  polyacetylene compounds from ginseng.

We now report the isolation of faltarinol and a new polyacetylene compound from fresh ginseng roots and their structural determination.

### RESULTS AND DISCUSSION

Four polyacetylenes, A-1–A-3, and B-1, were obtained and each component was examined by UV spectroscopy and compared with published data [6]. The spectra of the separated components showed typical polyacetylene vibrational bands (Table 1). They can be classified into two main polyacetylene groups having two different characteristic UV chromophores, namely a conjugated diyne system for A-1 and B-1 and a conjugated diyne-ene system for A-2 and A-3. Since A-2 and A-3 have higher molar absorption coefficients but show a much smaller absorption peak intensity by HPLC, they exist in minor quantities compared to those of A-1 and B-1. The preparative isolation and physical characterization of A-2 and A-3 were, therefore, not carried out. A-1 was identical to faltarinol by physical characterization.

The molecular skeleton of B-1 was easily recognized by  $^{13}C$  NMR (25.16 MHz,  $CDCl_3$ ) (Table 2). The proton wide band decoupled spectrum of B-1 showed the typical aliphatic methylene carbons at  $\delta$  22.2, 31.3, 28.8, 29.0, 26.0 and 27.1, the terminal methyl carbon of the straight aliphatic chain at  $\delta$  13.5, the two carbons of the terminal vinyl group at  $\delta$  117.3 and 136.3, the allylic carbon to the terminal vinyl group at  $\delta$  63.3, and the methylene carbon strongly shielded by conjugated triple bonds at  $\delta$  18.9. The resonance of quarternary carbons in the conjugated triple bonds exhibited at  $\delta$  77.6, 76.7 and 75.2 in B-1 were not definitely assigned because only three are observed instead of four, one being superimposed with solvent peaks. The

$^{13}C$  chemical shifts of two carbons in B-1, are shifted upfield from  $\delta$  126.6 and 122.1 to  $\delta$  56.5 and 54.0 as compared to A-1.

The IR of B-1 showed a hydroxyl group at  $3400\text{ cm}^{-1}$ , a methylene group at  $2940$  and  $2863\text{ cm}^{-1}$ , conjugated triple bonds at  $2260\text{ cm}^{-1}$ , C–O stretching of a secondary hydroxyl group at  $1120\text{ cm}^{-1}$ , and a terminal vinyl group at  $1000\text{--}900\text{ cm}^{-1}$ . Two differences between the spectra of A-1 and B-1 were observed, one being the difference in relative peak intensity at  $1120\text{ cm}^{-1}$ , the other being the absence of a peak indicating a *cis* internal double bond at  $690\text{ cm}^{-1}$  in spectrum of B-1. The reason for the former is attributed to the presence of additional secondary hydroxyl groups in B-1.

The  $^1H$  NMR spectra (79.542 MHz,  $CDCl_3$ ) of B-1 showed a complex spin system of a terminal vinyl group at  $\delta$  5.13–6.10, allylic protons to a terminal vinyl group at  $\delta$  4.79–4.92, methylene protons of straight hydrocarbon chains at  $\delta$  1.28–1.44 and corresponding terminal methyl group protons at  $\delta$  0.85. The absence of peaks in the range

Table 1. UV  $\lambda_{\text{max}}^{\text{hexane}}$  of ginseng polyacetylenes

Component	$\lambda_{\text{max}}$ (nm)	Band spacing ( $\text{cm}^{-1}$ )
A-1	257	2411
	242	1968
	231	2165
	220	—
B-1	254	2297
	240	2193
	228	2223
	217	—
A-2 and A-3	279	2181
	263	1977
	250	2194
	237	2054
	226	3371
	210	—

Table 2.  $^{13}\text{C}$ NMR chemical shifts of polyacetylene B-1

Carbon No.	Chemical shifts ( $\delta$ )
1	117.3
2	136.3
3	63.3
4	77.6
5	76.7
6	75.2
7	—
8	18.9
9	54.0
10	56.5
11	27.1
12	26.0
13	29.0
14	28.8
15	31.3
16	22.2
17	13.5

$\delta$  5.37–5.48 indicates the absence of protons bound to the internal double bond of the aliphatic hydrocarbon chain in contrast to falcarinol. The presence of additional secondary hydroxyl groups in B-1 as compared with A-1 may be responsible for the very complex spin system between  $\delta$  1.98 and 3.19 in B-1.

The mass spectra of B-1 determined by EI showed neither an  $\text{M}^+$  at  $m/z$  278 nor typical fragment peaks, probably due to its inherent instability. However, the elemental analysis data of B-1 are consistent with the molecular formula  $\text{C}_{17}\text{H}_{26}\text{O}_3$ .

From these experimental results, B-1 is identified as a  $\text{C}_{17}$  acetylenic compound differing only in two secondary hydroxyl groups at the 9 and 10 positions instead of the *cis* double bond in falcarinol. B-1 is, thus, heptadeca-1-ene-4,6-diyne-3,9,10-triol. This is a new acetylenic compound among the  $\text{C}_{17}$  naturally occurring polyacetylenes in Korean ginseng roots. The stereochemistry of the *vic*-diol group is being further pursued.

## EXPERIMENTAL

*Isolation of heptadeca-1-ene-4,6-diyne-3,9,10-triol.* Fresh Korean ginseng roots (*P. ginseng*, A. C. Meyer) 8 kg were finely crushed and extracted with MeOH. MeOH extracts were partitioned between petrol–MeOH (1:1). The petrol layer was washed with 5% NaOH soln several times and the solvent evaporated. The crude oily mixture obtained (4.5 g) was dissolved in the CC solvent (petrol and  $\text{Et}_2\text{O}$ ). Stepwise gradient elution from Si gel with petrol– $\text{Et}_2\text{O}$  (from 5:1 to 2:1) gave two main fractions A (ca 120 mg) and B (ca 330 mg) containing the polyacetylene compounds. Each fraction separated by CC was further chromatographed by HPLC. To monitor the polyacetylenes from A and B analytical liquid chromatography was performed under the following conditions; column:  $\mu$ -Bondapak CN (3.9 mm i.d.  $\times$  30 cm), solvents: *n*-hexane– $\text{CH}_2\text{Cl}_2$  (20:1) for fraction A; *n*-hexane– $\text{Et}_2\text{O}$  (20:1) for fraction B, flow rate: 1.0 ml/min, detector: UV (254 nm). The  $R_f$  values (min) were A-1 11.2, A-2 12.0, A-3 14.4 and B-1 8.3.

To isolate the polyacetylenes from each fraction, semi-prep. HPLC was carried out with the same instrument and the same microparticulate column. The polyacetylenes were collected in bottles immersed in dry ice– $\text{Me}_2\text{CO}$  and covered with Al foil to exclude light. The purity of each separated fraction was rechecked by analytical HPLC. For spectroscopic measurements, the collected fractions were concd by evaporation in a stream of  $\text{N}_2$ . Residual solvents were removed by rotary vacuum evaporator.

Analysis for B-1 requires for  $\text{C}_{17}\text{H}_{26}\text{O}_3$ : C, 73.38; H, 9.35; O, 17.27; found: C, 73.06; H, 9.40; O, 17.54%.

The wts of A-1 and B-1 recovered after chromatographic fractionation were 7 and 12 mg, respectively.

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